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# BS8161 PHYSICS AND CHEMISTRY LABORATORY L T P C

(Common to all branches of B.E. / B.Tech Programmes) 0

#### **OBJECTIVES:**

 To introduce different experiments to test basic understanding of physics concepts applied in optics, thermal physics, properties of matter and liquids.

## LIST OF EXPERIMENTS: PHYSICS LABORATORY (Any 5 Experiments)

- 1. Determination of rigidity modulus Torsion pendulum
- 2. Determination of Young's modulus by non-uniform bending method
- 3. (a) Determination of wavelength, and particle size using Laser(b) Determination of acceptance angle in an optical fiber.
- 4. Determination of thermal conductivity of a bad conductor Lee's Disc method.
- 5. Determination of velocity of sound and compressibility of liquid Ultrasonic interferometer
- 6. Determination of wavelength of mercury spectrum spectrometer grating
- 7. Determination of band gap of a semiconductor
- 8. Determination of thickness of a thin wire Air wedge method

## OUTCOMES:

Upon completion of the course, the students will be able to

• apply principles of elasticity, optics and thermal properties for engineering applications.

## CHEMISTRY LABORATORY: (Any seven experiments to be conducted)

## **OBJECTIVES**:

- To make the student to acquire practical skills in the determination of water quality parameters through volumetric and instrumental analysis.
- To acquaint the students with the determination of molecular weight of a polymer by viscometery.
  - 1. Estimation of HCl using Na<sub>2</sub>CO<sub>3</sub> as primary standard and Determination of alkalinity in water sample.
  - 2. Determination of total, temporary & permanent hardness of water by EDTA method.
  - 3. Determination of DO content of water sample by Winkler's method.
  - 4. Determination of chloride content of water sample by argentometric method.
  - 5. Estimation of copper content of the given solution by lodometry.
  - 6. Determination of strength of given hydrochloric acid using pH meter.
  - 7. Determination of strength of acids in a mixture of acids using conductivity meter.
  - 8. Estimation of iron content of the given solution using potentiometer.
  - 9. Estimation of iron content of the water sample using spectrophotometer (1, 10-Phenanthroline / thiocyanate method).
  - 10. Estimation of sodium and potassium present in water using flame photometer.
  - 11. Determination of molecular weight of polyvinyl alcohol using Ostwald viscometer.
  - 12. Pseudo first order kinetics-ester hydrolysis.
  - 13. Corrosion experiment-weight loss method.
  - 14. Determination of CMC.
  - 15. Phase change in a solid.
  - 16. Conductometric titration of strong acid vs strong base.

## **OUTCOMES:**

The students will be outfitted with hands-on knowledge in the quantitative chemical analysis of water quality related parameters.

## TEXTBOOKS:

- TOTAL: 30 PERIODS
- 1. Vogel's Textbook of Quantitative Chemical Analysis (8<sup>TH</sup> edition, 2014)

## TOTAL: 30 PERIODS

2